## Summative Assessment II (March- 2017) Marking Scheme Class X – Outside Delhi

Code No. 31/1

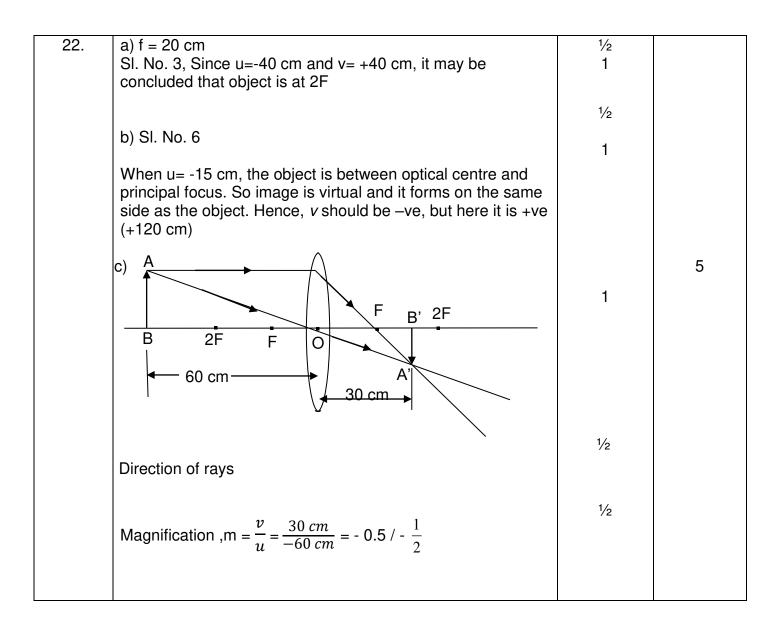
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Q. No.	Expected Answer / Value point Section - A	Distribution of marks	Marks
1.	C <sub>2</sub> H <sub>6</sub> , C <sub>3</sub> H <sub>8</sub>	1/2 + 1/2	1
2.	Creation of DNA copy / Replication / Copying of DNA	1	1
3.	1000000 J	1	1
4.	<ul> <li>Virtual</li> <li>Erect</li> <li>Diminished</li> <li>On the same side as the object / or any other characteristic</li> </ul>	½ x 4	2
5.	<ul> <li>Conserving forests helps in (i) retaining sub soil water and (ii) checking floods / any other</li> <li>Conserving wild life helps in (i) maintaining ecological balance and (ii) protecting the nature (or any other)</li> </ul>	½ x 4	2
6.	<ul> <li>Water stored during rainy season can be used as and when required by the community.</li> <li>Ground water level increases due to recharging.</li> </ul>	1 + 1	2
7.	Н Н     Н—С—С—ОН	1	
	́н́н́	1⁄2	
	Ethene is produced	1	
	$CH_{3}CH_{2}OH \xrightarrow{Conc. H_{2}SO_{4}} C_{2}H_{4} + H_{2}O$ $443K$	1/2	3
	Conc. $H_2SO_4$ acts as a dehydrating agent.		
8.	Esterification – A process in which an alcohol and a carboxylic acid react in the presence of conc. $H_2SO_4$ to form an ester.	1⁄2	
	$CH_{3}COOH + C_{2}H_{5}OH \xrightarrow{Conc. H_{2}SO_{4}} \rightarrow CH_{3}COOC_{2}H_{5} + H_{2}O$	1⁄2	

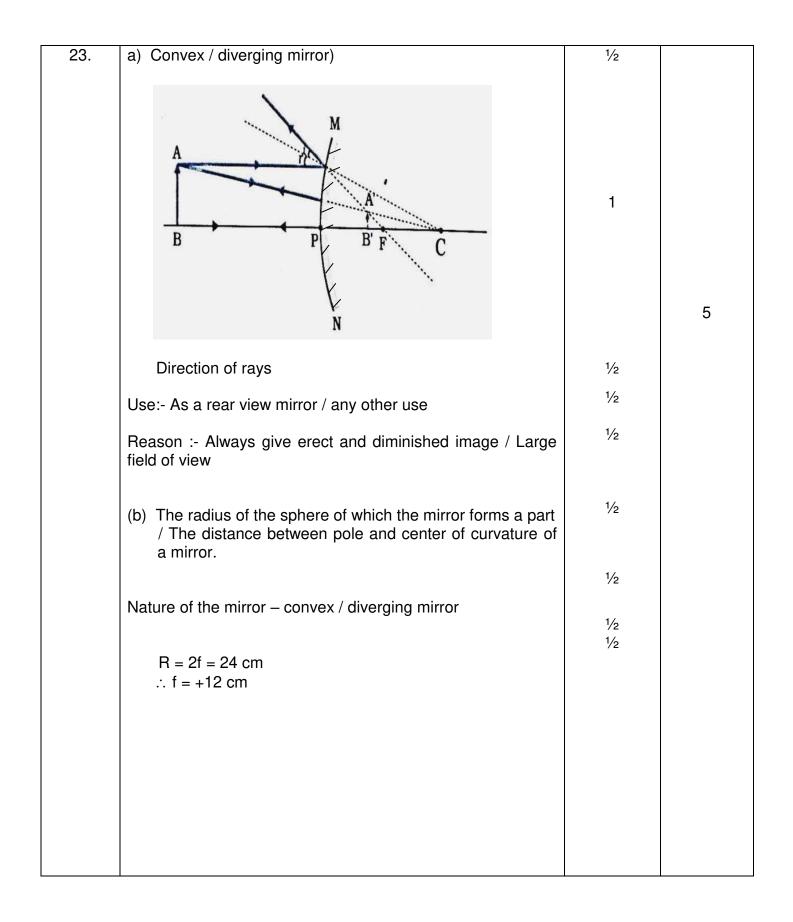
	<ul> <li>Saponification – A process in which an ester reacts with sodium hydroxide to form sodium salt of an acid and alcohol / an ester reacts in the presence of an acid or a base to give back the alcohol and carboxylic acid.</li> </ul>	1⁄2	
	<ul> <li>CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub> + NaOH → C<sub>2</sub>H<sub>5</sub>OH + CH<sub>3</sub>COONa</li> </ul>	1⁄2	3
	<ul> <li>Esters are used in ice creams / perfumes</li> <li>Saponification process is used in preparation of soap.</li> </ul>	1/2 1/2	
9.	<ul> <li>Periods – 7, Groups – 18</li> </ul>	1⁄2 + 1⁄2	
	• Metallic character decreases along the period because effective nuclear charge increases on the valence electrons hence decrease in tendency to lose electrons.	1⁄2 + 1⁄2	3
	<ul> <li>Metallic character increases down a group because effective nuclear charge experienced by valence electrons decrease, hence tendency to lose electron decreases.</li> </ul>	1⁄2 + 1⁄2	
10.	<ul> <li>Aluminium (Al) Reason – Valency of Na is 1, Mg is 2, Al is 3</li> </ul>	1/2 1/2	
	<ul> <li>Sodium (Na) Reason – As we move from left to right in a period, the atomic radius decreases / increase in nuclear charge pulls the electrons closer to the nucleus reducing the atomic size.</li> </ul>	1/2 1/2	
	<ul> <li>Sodium (Na) Reason – Reactivity decreases on moving from left to right in a period / any other reason</li> </ul>	1/2 1/2	3
11.	<ul> <li>For continuation of species / perpetuation of species</li> <li>It promotes diversity of characters / helps to show the variations which enhances the survival chances.</li> </ul>	1 1	
	<ul> <li>Increases population of a species / any other answer</li> </ul>	1	3
12.	<ul> <li>Vegetative propagation – A process in which any vegetative part of a plant (root, stem or leaf) gives rise to a new plant under appropriate conditions.</li> </ul>	1	
	<ul> <li>Two advantages :- (i) Large number of plants obtained in a short interval. (ii) Propagation of seedless plants is made possible / any other advantage.</li> </ul>	1/2 1/2	
	<ul> <li>Two disadvantages :- (i) No genetic variations, so, less</li> <li>adaptability to the environment. (ii) The disease of plants gets</li> </ul>	1/2 1/2	3
13.	<ul> <li>transferred to the offsprings.</li> <li>Three techniques – Barrier method, chemical method, surgical</li> </ul>	½ x 3	5
	<ul><li>method</li><li>Chemical method</li></ul>	1/2	
	<ul> <li>It maintains health of the woman, parents can provide more attention to children / more resources are available to the family</li> </ul>	1⁄2 x 2	
	/ any other.		3

14.	<ul> <li>In Mendel's experiment, when pure tall pea plants were cr with pure dwarf pea plants, only tall pea plants were obtain F<sub>1</sub> generation.</li> </ul>		1	
	<ul> <li>On selfing the pea plants of F1 generation both tall and dw pea plants were obtained in F2 generation.</li> </ul>	varf	1	
	<ul> <li>Reappearance of the dwarf pea plants in F<sub>2</sub> generation pro that the dwarf trait was inherited but not expressed in F<sub>1</sub> generation.</li> </ul>	oves	1	
	Note:- If explained with flow chart with proper description, full r be awarded.	marks		3
15.	<ul> <li>Different life forms have evolved during the course of evo Classification deals with the grouping of these life forms in groups and sub groups based on similarities and differenc</li> <li>The more characteristics any two species have in commor more closely they are related.</li> <li>Thus classification helps in tracing the evolutionary relation between the two organisms. Hence, evolution and classific are interlinked.</li> </ul>	to es. n, nship	1 1 1	3
16.	Concave / diverging lens. A $2F_1$ B $F_1$ B N	1/2 1		3
	Direction of rays $f = \frac{1}{P}$ , P = -10D,	1/2 1/2		
	$f = \frac{1}{-10D} = -0.1 \text{ m/-10 cm}$	1⁄2		

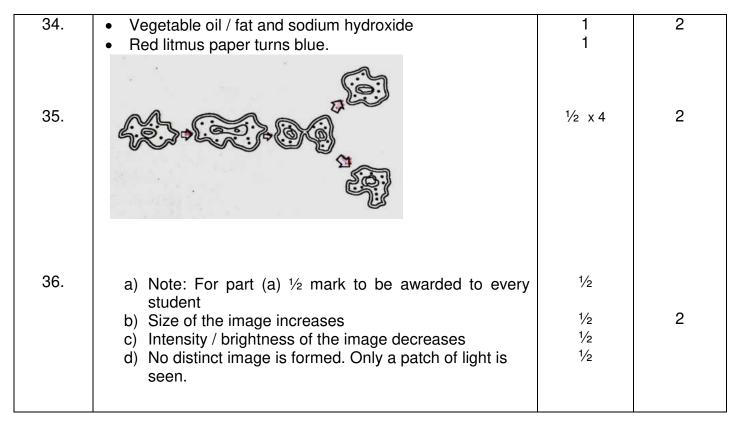
17.	Different colours of light bend through different angles with respect to the incident ray / different speed of different colours of light in glass / different values of refractive index of glass for different colours of light.	1	3
	White light R R R White light P <sub>1</sub> A	1	
	Direction of ray & labelling	1/2 + 1/2	
18.	<ul> <li>a) Two ways of creating awareness</li> <li>Door to door campaigning</li> <li>Nukkad natak / any other method.</li> </ul>	1 + 1	3
	b) Rain water harvesting with explanation / preventing over extraction of underground water / any other method	1⁄2 + 1⁄2	
19.	<ul> <li>Compounds of hydrogen and carbon</li> <li>Alkanes - CnH<sub>2n+2</sub></li> <li>Alkenes - CnH<sub>2n</sub></li> <li>Alkynes - CnH<sub>2n-2</sub></li> </ul>	1/2 1/2 1/2 1/2 1/2	
	• . H = H = H = H H = C = C = H, $H = C = C = HH = H$	½ x3	5
	Addition reaction / hydrogenation	1⁄2	
	$R \xrightarrow{C} C = C \xrightarrow{R} \xrightarrow{R \xrightarrow{Ni/Pd catalyst}}_{H_2} H \xrightarrow{R} \xrightarrow{R} \stackrel{R}{\downarrow} \stackrel{R}{\downarrow}_{I}$ $R \xrightarrow{R} \xrightarrow{R} H \xrightarrow{R} \xrightarrow{R} H \xrightarrow{R} H$	1	

20.	(a) Functions :		]
	(I) Ovary:-		
	(i) Production of female hormones / estrogen / progesterone	$\frac{1}{2} + \frac{1}{2}$	
	(ii) Production of female gamete/egg/ germ cells	72 + 72	
	(II) Uterus:-     (i) Implementation of averate / embryo		
	<ul><li>(i) Implantation of zygote / embryo</li><li>(ii) Nourishment of developing embryo</li></ul>	1/2 + 1/2	5
	(III) Fallopian tube :-     (i) Transfer of female gemete from the overv		
	<ul><li>(i) Transfer of female gamete from the ovary</li><li>(ii) Site of fertilisation</li></ul>	<sup>1</sup> /2 + <sup>1</sup> /2	
	(b) Structure of placenta :- It is a special disc like tissue embedded in mother's uterine wall and connected to the	1	
	foetus / embryo.	I	
	Functions of placenta :- It provides a large surface area for glucose and oxygen / nutrients to pass from mother's body		
	to the developing / developed embryo / foetus and also	1	
	helps in passing the waste from the foetus / embryo to the		
	mother's body.		
21.	Acquired traits – Traits which develop in the life time of	1	
	an individual and do not pass to the progeny.	17	
	Example- Learning a skill such as dance / music / loss of body parts / weight / any other example.	1/2	
	<ul> <li>Inherited traits – Traits present in the gamete / germ</li> </ul>	1	
	cells which can be seen in the progeny.	47	
	Example – Skin colour / eyebrows / any other example.	1/2	5
	Reasons – Traits / characteristics acquired during one's		0
	life time do not bring any change in the DNA of the	1	
	reproducing cells / germ cells.		
	Examples - Decrease in body weight of beetles due to starvation do not pass on to the next generation	1	
	because there is no change in the germ cells of beetles.		





24.	2)		
24.	<ul> <li>a)</li> <li>excessive curvature of the eye lens</li> <li>elongation of the eyeball</li> </ul>	1/2 1/2	
	i)		
	Parallel rays from distant object (at infinity) Image is formed in front of the retina	1	
	ii) Concave / diverging lens	1/2	5
	Parallel rays from distant object (at infinity) F Virtual image formed Concave lens to correct myopia	1	0
	b) $f = -5$ m (since lens is concave)	1/2	
		1/2	
	$P = \frac{1}{f(\text{metre})}$ P= - 0.2 D	1/2	
25.	D Section - B	1	
26.	С	1	
27.	D	1	
28.	В	1	
29.	С	1	
30.	D	1	
31.	D	1	
32.	В	1	
33.	С	1	9



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